THE SYNTHESES OF 1-N-[(S)-4-AMINO-2-HYDROXYBUTYRYL]GENTAMICIN C₁ AND 1-N-[(S)-3-AMINO-2-HYDROXYPROPIONYL]GENTAMICIN C₁

Sir:

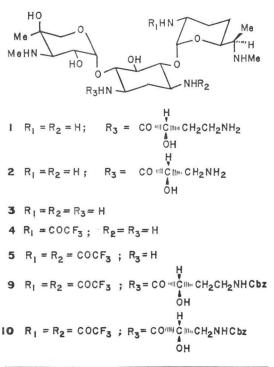
The naturally occurring aminoglycoside antibiotic butirosin has been shown to possess an N-(S)-4-amino-2-hydroxybutyryl [AHB] substituent¹⁾ which is uniquely associated with the activity of this antibiotic against species of Pseudomonas aeruginosa. KAWAGUCHI and coworkers²⁾ synthesised an analogous derivative of kanamycin A, 1-N-[(S)-4-amino-2-hydroxybutyryl] kanamycin A (amikacin), which not only possessed anti-Pseudomonas activity not shown by the parent antibiotic, but also exhibited marked inhibition of organisms resistant to kanamycin A and other aminoglycoside antibiotics.^{2,3)} Since this time analogous derivatives of a number of other aminoglycosides have been reported, for example, of 3',4'-dideoxykanamycin B.⁴⁾

In an investigation of structure-activity relationships in the butirosin series, HASKELL and coworkers⁵⁾ prepared semisynthetic analogs in which the structure of the aminohydroxyacyl side chain was varied. Of these compounds only the lower homolog, *i.e.* the 1-*N*-3-amino-2-hydroxypropionyl [AHP] derivative, showed activity comparable to butirosin *in vitro*. In this communication we report the synthesis and *in vitro* antibacterial activity of 1-N-[(S)-4-amino-2-hydroxybutyryl]gentamicinC₁ [AHB-C₁] (1) and <math>1-N-[(S)-3-amino-2-hydroxypropionyl]gentamicin C₁ [AHP-C₁] (2).*

The approach taken was similar to that used by previous workers^{2,4)} involving sequential selective blocking of amino groups; however, we have found it convenient to use trifluoroacetyl (TFA) protecting groups, introduced using ethyl trifluorothiolacetate. The order of reactivity of the amino functions of gentamicin C_1 (3) to this reagent was 2'>3>1>6' and 3''.

Treatment of a methanolic solution of 3 with one equivalent of ethyl trifluorothiolacetate at room temperature, followed by isolation of the major product by chromatography over silica gel, eluting with the lower phase of a 2:1:1 chloroform-methanol-3N ammonium hydroxide solvent system, gave 2'-N-trifluoroacetylgentamicin C_1 (4), $C_{23}H_{42}N_5O_8F_3\cdot H_2O^{\dagger}$, m.p. $108 \sim 111^{\circ}$ C, $[\alpha]_{D}^{26} + 128^{\circ}$ (c 0.4, H₂O), in 69% yield. The location of the TFA group was indicated by mass spectrometry.** The molecular ion and $[M+1]^+$ peaks of 4 appeared at m/e 573 and 574 with intense ions at m/e160, corresponding to an unsubstituted garosamine⁶⁾ fragment, m/e 253, consistent with a monotrifluoroacetyl purpurosamine A6) fragment and m/e 191, 173, 163 and 145 indicating

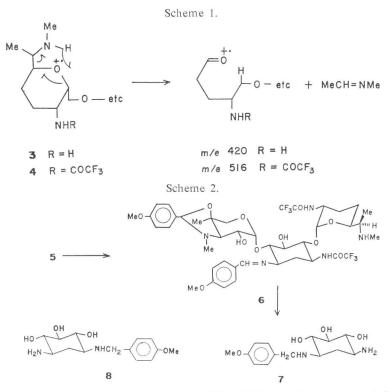




[†] Satisfactory elemental analyses were obtained for all compounds indicated by molecular formulae as shown.

^{*} Since this work was completed, compound 1 was reported in a patent (U.S. Patent 3,780,018, December 18, 1973). The preparation described therein involved reaction of gentamicin C_1 directly without protection of the 2' and 3 amino groups. Also since this work was completed, the preparation of 1-N-3-amino-2-hydroxypropionyl derivatives of some kanamycins has been reported (KONDO, S.; K. IINUMA, M. HAMADA, K. MAEDA & H. UMEZAWA: J. Antibiotics 27: 90~93, 1974).

^{**} Mass spectra were obtained using a Varian MAT CH5 spectrometer at 70 eV with a probe temperature of $200 \sim 250^{\circ}$ C. The direct inlet technique was used.

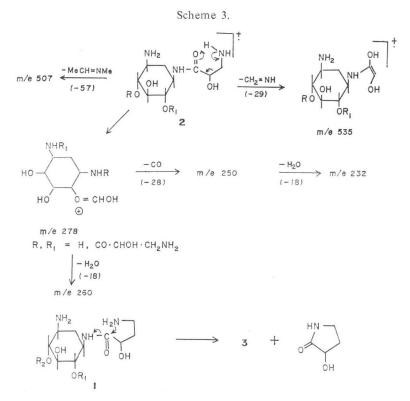


an unsubstituted 2-deoxystreptamine fragment.⁷⁾ A relatively intense $[M-57]^+$ ion at m/e 516 was consistent with loss of the purpurosamine A side chain, possibly as shown in Scheme 1, indicating the location of the TFA group at N-2'. An analogous ion is observed at m/e 420 in the mass spectrum of gentamicin C₁, and its composition in this latter case was confirmed by high resolution measurements. Other ions in the mass spectrum of 4, including the expected series of pseudodisaccharide ions, were consistent with the assigned structure.

Treatment of 4 with one equivalent of ethyl trifluorothiolacetate, followed by chromatography gave, in 42 % yield, 2',3-di-*N*-trifluoroacetylgentamicin C₁ (5), C₂₈H₄₁N₅O₉F₈·H₂O, m.p. 121~129°C, $[\alpha]_{D}^{26}$ +121° (c 0.3, H₂O). The second TFA group in 5 was located on the 2-deoxystreptamine moiety by mass spectrometry. The series of ions identified with the 2-deoxystreptamine residue were shifted by 96 mass units appearing at *m/e* 287, 269, 259 and 241, whereas the monosaccharide ions were unshifted at *m/e* 160 and 253. The molecular ion and [M+1]⁺ peak appeared at *m/e* 669 and 670 and the [M-MeCH=NMe]⁺⁻ ion at *m/e*

612. Differentiation between 1-N and 3-N substitution was made on the basis of experiments outlined in Scheme 2. Condensation of 5 with p-methoxybenzaldehyde gave the bis-pmethoxybenzylidene derivative 6 (M⁺⁻ at m/e905) which was reduced with sodium borohydride, hydrolysed with 6 N hydrochloric acid and the product chromatographed to give a 74 % yield of 2-deoxy-1-N-p-methoxybenzyl-Dstreptamine (7), $[\alpha]_{D}^{26}$ -44.6° (c 0.5, H₂O). This compound was identical spectroscopically and by chromatographic criteria with an authentic sample of the enantiomeric 3-N-p-methoxybenzyl derivative $8^{(2)} [\alpha]_{\rm D} + 44^{\circ} (c \ 0.3, \ {\rm H}_2{\rm O}),$ but had the opposite sign of optical rotation.* Compound 5 was also prepared directly from 3 in 34 % yield by reaction with two equivalents of ethyl trifluorothiolacetate. Condensation of 5 with one equivalent of ethyl trifluorothiolacetate resulted in reaction at N-1 affording 1,3,2'-tri-N-trifluoroacetylgentamicin C_1 , $C_{27}H_{40}N_5O_{10}F_9 \cdot H_2O$, m.p. 147 ~ 151°C, $[\alpha]_{\rm D}^{26}$ +114° (c 0.38, H₂O), mass spectral peaks at m/e 160, 253 (intense), m/e 383, 365, 355,

^{*} We thank Dr. I. R. HOOPER of Bristol Laboratories, Syracuse, N.Y. for an authentic sample of this material.



337 (weak).

Condensation of 5 with 1.2 equivalents of N-[(S)-4-benzyloxycarbonylamino-2-hydroxybutyryloxy]succinimide²⁾ at room temperature in tetrahydrofuran solution afforded, after purification by chromatography over silica gel, the 1 - N - [(S) - 4 - benzyloxycarbonylamino - 2 hydroxybutyryl] derivative 9, $C_{37}H_{54}N_6O_{18}F_6$, m.p. 115 \sim 120°C, $[\alpha]_{D}^{28}$ +80° (c 0.35, MeOH) in 51% yield. Removal of the trifluoroacetyl groups was effected by stirring 9 with 5 N ammonium hydroxide in aqueous methanol at room temperature for 3 days with TLC monitoring. The resulting product was hydrogenated in acetic acid over a 10 % palladium on carbon catalyst to remove the carbobenzyloxy group affording the desired product. Purification was effected by chromatography over silica gel, eluting with the lower phase of a 1:1:1 chloroform - methanol - conc. ammonium hydroxide solvent system affording 1-N-[(S)-4-amino-2-hydroxybutyryl] gentamicin C_1 (1), $C_{25}H_{50}N_6O_6 \cdot 2H_9O_7$ in 78 % yield from 9. Data for 1 were as follows: m.p. $103 \sim 110^{\circ}$ C, $[\alpha]_{D}^{26}$ $+101^{\circ}$ (c 0.45, H₂O), PMR (100 MHz, D₂O) δ 1.01 (3H, d, J=7 Hz, CH₃-CH), 1.17 (3H, s,

<u>CH₃-C)</u>, 2.32 (3H, s, <u>CH₃-N)</u>, 2.47 (3H, s, <u>CH₃-N)</u>, 4.19 (1H, t, J=4.2 Hz, H-2^{'''}), 5.08 (1H, d, J=4.5 Hz, H-1^{''}), 5.13 (1H, d, J=4 Hz, H-1[']) ppm. Irradiation at ∂ 1.73 ppm collapsed the triplet at ∂ 4.19 ppm to a singlet. The structure of 1 was confirmed by the method of KAWAGUCH1²⁾. Condensation of 1 with *p*-methoxybenzaldehyde afforded the tetra-*N p*-methoxybenzylidene derivative, which was reduced with sodium borohydride, and hydrolysed with acid affording 2-deoxy-3-*N*-*p*methoxybenzyl-D-streptamine (8).

In a similar manner, condensation of **5** with N-[(S)-3-benzyloxycarbonylamino-2-hydroxypropionyloxy]succinimide afforded the 1-Nsubstituted compound **10**, $C_{36}H_{52}N_6O_{13}F_6 \cdot H_2O$, m.p. 125~131°C, $[\alpha]_D^{23}+93°$ (c 0.34, MeOH). Removal of the protecting groups as described previously gave 1-N-[(S)-3-amino-2-hydroxypropionyl]gentamicin C_1 (**2**), $C_{24}H_{48}N_6O_9 \cdot H_2O$, m.p. 109~119°C, $[\alpha]_D^{28}+98°$ (c 0.27, H_2O), PMR (100 MHz, D_2O) δ 1.10 (3H, d, J=7 Hz, CH₃-CH), 1.19 (3H, s, CH₃-C), 2.41 (3H, s, CH₃-N), 2.49 (3H, s, CH₃-N), ca 4.16 (2H, overlapping signals. H-2''' and H-5''₆(α), 5.08 (1H, d, J=4 Hz, H-1''), 5.18 (1H, d, J=3.5

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	<i>itro</i> antibacterial activities of 1-N-[(S)]-4-amino-2-hydroxybutyryl]gentamicin
C_1 (AH	$HB-C_1$, $1-N-[(S)-3-amino-2-hydroxypropionyl]gentamicin C_1 (AHP-C_1)$
	and gentamicin C_1 (C_1)

Test orga	st organism	Resistance mechanism	Minimal inhibitor concentration (mcg/ml)*		
i est organism			C_1	AHB-C ₁	AHP-C
S. aureus	FDA 209 P		0.03	0.3	0.3
"	Wood		0.03	0.3	0.3
S. pyogen	es C		3.0	>25	17.5
//	27		3.0	7.5	>25
//	A Alvarez		0.75	17.5	
<i>E. coli</i> LA290/R55		2"-O-adenylylating	>25	0.3	0.75
" W677/R55		"	>25	0.75	0.3
"JR	.88	3-N-acetylating	>25	0.75	0.75
" Ba	iker 2		0.75	0.75	3.0
<i>"</i> F	14-BK		0.75	0.75	0.75
" St. M. 589			3.0	3.0	0.75
" R5/W677		6'-N-acetylating	0.3	0.3	0.3
P. aeruginosa D2			3.0	17.5	17.5
//	NRRL 3223		0.75	3.0	3.0
11	1395		0.75	7.5	3.0
//	Travers	3-N-acetylating	>25	7.5	7.5
"	Stone 138	"	>25	7.5	3.0
"	Shriners 10099	Unknown	>25	>25	>25
11	GN 315	6'-N-acetylating	3.0	3.0	3.0
K. pneumoniae Ad 17			0.3	0.3	0.075
"	Ad 18		0.3	0.3	0.3
//	Georgetown	2"-O-adenylylating	>25	0.75	0.3
Providence 164		2'N-acetylating	>25	>25	>25
P. mirabilis Harding			3.0	0.3	0.75
P. rettgeri Membel			3.0	3.0	7.5
B. subtilis	6623		0.03	0.03	0.03

* In MUELLER-HINTON broth pH 7.2.

Hz, H-1') ppm. The mass spectrum of 2 gave relatively intense ions at $m/e 565 (M+1)^+$, 535 and 507; an intense series of ions at m/e 278, 260, 250 and 232 corresponded to the normal series of deoxystreptamine ions⁷ displaced to higher mass by the additional substituent (see Scheme 3). The mass spectrum of 1 on the other hand was essentially the same as that of the parent gentamicin C₁ except for a weak ion at $m/e 560 [M-H_2O]^+$. These date suggest that compound 1 undergoes thermal elimination of the side chain in the mass spectrometer by the cyclic mechanism shown (Scheme 3). Such a mechanism would not be easily available to the lower homolog 2.

The in vitro antibacterial activities of 1

[AHB-C₁] and 2 [AHP-C₁] against some representative organisms are shown in the Table in comparison to gentamicin C₁. In general, 1 and 2 have excellent activity and are potent inhibitors of organisms which inactivate gentamicin C₁ by 2"-O-adenylylation and 3-N-acety-lation. They are however inactive against a strain of *Providence* which inactivates gentamicin by 2'-N-acetylation.⁸⁾ 1 and 2 have only weak activity against *S. pyogenes* strains and are relatively weak against gentamicin-sensitive *Pseudomonas* compared to gentamicin itself.

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